

Lewis Structure, Molecular Shape, and Polarity Practice

SO_4^{2-} VSEPR shape = _____ Polar? Y N	CBr_4 VSEPR shape = _____ Polar? Y N
PBr_3 VSEPR shape = _____ Polar? Y N	CH_3Br VSEPR shape = _____ Polar? Y N
SF_4 VSEPR shape = _____ Polar? Y N	H_2O VSEPR shape = _____ Polar? Y N
CO VSEPR shape = _____ Polar? Y N	SF_6 VSEPR shape = _____ Polar? Y N
CO_2 VSEPR shape = _____ Polar? Y N	PF_5 VSEPR shape = _____ Polar? Y N
NH_4^{1+} VSEPR shape = _____ Polar? Y N	NH_3 VSEPR shape = _____ Polar? Y N

NH_2^{1-} VSEPR shape = _____ Polar? Y N	ClO_3^{1-} VSEPR shape = _____ Polar? Y N
ClF_4^{1+} VSEPR shape = _____ Polar? Y N	I_3^{1-} VSEPR shape = _____ Polar? Y N
BrF_3 VSEPR shape = _____ Polar? Y N	NO_3^{1-} VSEPR shape = _____ Polar? Y N
BF_3 VSEPR shape = _____ Polar? Y N	BeF_2 VSEPR shape = _____ Polar? Y N
IBr_5 VSEPR shape = _____ Polar? Y N	XeF_4 VSEPR shape = _____ Polar? Y N
HClO_4 VSEPR shape = _____ Polar? Y N	SO_2 VSEPR shape = _____ Polar? Y N