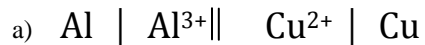
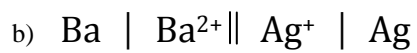


Chemistry
Electrochemical cell practice

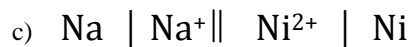
Calculate E°_{cell} and write the half reactions and the overall cell reaction for the following:



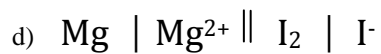
$E^{\circ}_{\text{cell}} = \underline{\hspace{2cm}}$



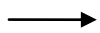
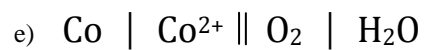
$E^{\circ}_{\text{cell}} = \underline{\hspace{2cm}}$



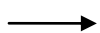
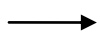
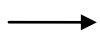
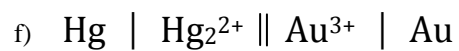
$E^{\circ}_{\text{cell}} = \underline{\hspace{2cm}}$



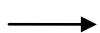
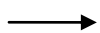
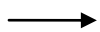
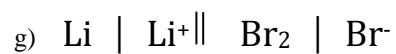
$E^{\circ}_{\text{cell}} = \underline{\hspace{2cm}}$



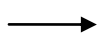
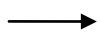
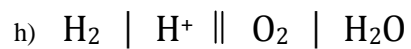
$E^\circ_{\text{cell}} = \underline{\hspace{2cm}}$



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