

DOUBLE REPLACEMENT REACTION PRACTICE

Given the following five reactants, write balanced molecular equations for each possible combination of two reactants. Determine if there would be a reaction. If there is a reaction, write the balanced ionic and net ionic equation for the reaction.

Given 0.1M solutions of: Na_3PO_4 , K_2CO_3 , $\text{Pb}(\text{NO}_3)_2$, $\text{Ba}(\text{OH})_2$, and AlCl_3 .

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